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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$ Classification Of Chemical Reactions Reactions, Much Like Elements And Compounds, Can Be Classified According To Type. The Ability To Recognize And Classify Reactions Can Help Us Predict Apr 15th, 2024

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Cannot Hav May 20th, 2024

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Type Decomposition $6 \text{ H}_2\text{O} \rightarrow 2 \text{H}_2 + \text{O}_2$,
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Reactions Into Different Types Such As A Synthesis
Combination Decomposition Single-replacement
Double-replacement With Feb 1th, 2024

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Year 10 Chemistry Topic 4 Reactions FACT SHEET
Reactions 1. What Do We Call Chemicals Present At
The Start Of A Reaction? Reactants 2. What Do We Call
Chemicals Made In A Reaction? Products 3. What Is
The State Symbol For A Solid? (s) 4. What Is The State
Symbol For A Liquid? (l) 5. What Is The State Symbol
For A Gas? (g) 6. Mar 3th, 2024

Chapter 7 Chemical Reactions Section 7.1 Describing Reactions

Measure Amounts Of A Substance Because Chemical
Reactions Often Involve Large Numbers Of Small

Particles. 10. Circle The Letter Of The Correct Answer.
If One Carbon Atom Has An Atomic Mass Of 12.0 Amu
And One Oxygen Atom Has An Atomic Mass Of 16.0
Amu, What Is The Molar Mass Of Carbon Dioxide?
A.28.0 Amub.44.0 Amu C.28.0 Gd.44.0 G 11. Jan 6th,
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