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Thermochemistry

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Student Worksheet For Thermochemistry

Bonds), But The Answer Would Be The Same Regardless. You Only Need To Account For Bonds That Participate In The Reaction. 9. Bond Energy Per Bond (kJ Per Bond) H-Cl 427 S-Cl 271 H-S 347 Cl-Cl 243 2 HCl + SCl 2 H 2 S + 2 Cl 2 Reactants Bond Number Of Bonds/Coefficient Bond Energy (each) Total Energy HCl 2 427 854 SCl 2 271 542 Total 1396 Feb 8th, 2024

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11.3: Heat In Changes Of State Enthalpy (H): - The Amount Of Internal Energy At A Specific Pressure And Volume. Energy Involved In Physical Change (Temperature Or Phase Change): Heating Curve: - A Graph Of Temeparture Versus Time As A Substance Is Heated From A Solid Phase To A Gaseous Phase. Jan 5th, 2024

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