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Combusted In A Limited Supply Of Pure Oxygen. 0.50g Of The Carbon Combusts To
Form CO₂ And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture
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Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline
Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play
A Role, Since It Is In Excess. Then, N (N) 0.0322 .aniline N Aceta II. Ide Mol We Find
The Molar Mass Of Acetanilide: (N) () 12 8 1 9 14 1 16 1 135(). 89 M Aceta ... 2th,
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