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Equation For The Reaction And Either The Masses Of Solid Reactants And Products  
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That This Number Of Moles Of SO<sub>3</sub> Has. The First Step Resembles The Exercises We  
Did In Section 6.4 "Mole-Mole Relationships In Chemical Reactions". As Usual, We  
Start With The Quantity We Were Given: 3.59 Mol Fe<sub>2</sub>O<sub>3</sub> × 3 Mol SO<sub>3</sub> / 1 Mol Fe  
2O<sub>3</sub> = 10.77 Mol SO<sub>3</sub> The Mol Fe<sub>2</sub>O<sub>3</sub> Units Cancel, Leaving Mol SO<sub>3</sub> ... Apr 17th,

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