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Equation! $4\text{Al} + 3\text{O}_2$ Pg. 277, #3 - "Given" And An "Unknown" - Pg. 296 - In-class Problems 10a, 11a Feb 1th, 2024

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PRACTICE PACKET: Unit 7 Moles & Stoichiometry 5 www.chempride.weebly.com
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(Show All Work) 1. MgO 5. $\text{Ca}(\text{OH})_2$ 2. NaHCO_3 6. CH_4 3. $\text{C}_6\text{H}_{12}\text{O}_6$ 7. NH_3 4.
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Chemistry 11 Stoichiometry Review Worksheet Name: KEY 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 \text{ O}_2 \times 1 \text{ mol } 32.00 \text{ g} \times 1 \text{ mol} = 3.11 \text{ g O}_2$ %Yield = $\frac{1.45 \text{ g } 3.11 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$ 34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%? g pure FeCO_3 Jun 4th, 2024

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