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1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At
The Formation Of Ammonia (NH_3) From Nitrogen And Hydrogen (see Section
Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The
Equation: $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ (1) $H_0 = 92 \text{ kJ/mol}$ $S_0 = 192 \text{ J/mol K}$ 4th, 2024 Chemical
Engineering, CM3510: Chemical Reaction ...Grading: Exam 1 15% Exam 2 15%
Exam 3 15% Final Exam 20% Research Paper 10% Homework 20% Class
Participation 5% Last Year, The Following Grading Scale Was Used: $A \geq 92$ $A/B \geq 89$
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