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CARBON AND ITS COMPOUNDS OCCURRENCE OF CARBON

Amorphous Carbon Is Black And Has The Lowest Density. It Is Non Crystalline And Fairly Conducts Electricity. Amorphous Carbon Exists In Several Forms Like Wood Charcoal, Animal Charcoal, Sugar Charcoal, Lamp Black And Coke. I) Wood Charcoal It Is Formed When Wood Is Heated In Limited Supply Of Air (destructive Distillation Of Wood). 8th, 2024

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SECTION 3-1 REVIEW CARBON COMPOUNDS

Modern Biology Study Guide SECTION 3-1 REVIEW CARBON COMPOUNDS VOCABULARY REVIEW Define The Following Terms And Provide One Example For Each. 1. Organic Compound 2. Functional Group 3. Alcohol 4. Monomer 5. Polymer MULTIPLE CHOICE Write The Correct Letter In The Blank. 1. Organi 19th, 2024

Section 2-3 2-3 Carbon Compounds - Weebly

With Compounds Called Fatty Acids, as Shown In Figure 2-14. If Each Carbon Atom In A Lipid's Fatty Acid Chains Is Joined To Another Carbon Atom By A Single Bond, the Lipid Is Said To Be Saturated. The Term Saturated Is Used Because The Fatty Acids Contain The Maximum Possible Number O 17th, 2024

Carbon Compounds (Section 2.3)

Carbohydrates Are Compounds Made Up Of Carbon, Hydrogen, And Oxygen Atoms, Usually In A Ratio Of 1 : 2 : 1. (C 6H 12O 6) Living Things Use Carbohydrates As Their Main Source Of Energy. The Breakdown Of Sugars, Such As Glucose, Supplies Immediate Energy For Cell Activities. Plants, Some An 10th, 2024

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Section 2-3 Carbon Compounds. (pages 44-48) This Section Explains How The Element Carbon Is Able To Form Millions Of Carbon, Or Organic, Compounds. It Also Describes The Four Groups Of Organic Compounds Found In Living Things. The Chemistry Of Carbon 8th, 2024

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Section 2-3 Carbon Compounds (pages 44-48) Key Concept • What Are The Functions Of Each Group Of Organic Compounds? The Chemistry Of Carbon (page 44) 1. How Many Valence Electrons Does Each Carbon Atom Have? 2. What Gives Carbon The Ability To Form Chains That Are Almost Unlimite 21th, 2024

Section 3 Carbon Compounds

2. Monosaccharides (simple Sugar) Have The Formula Where N Can Be . The Most Common And Important Monosaccharide Is _____, Which Is A Six-carbon, So Has The Formula C 6H 12O 6. Its Structure Is: A-glucose (used To Make Starch And Gl 7th, 2024

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