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16. What Is The Half Life In Sec For A First Order Reaction With  $k = 2.03 \times 10^{-3} \text{ s}^{-1}$ ? A) 323 B) 369 C) 341 D) 422 E) 395

17. The Half Life Of A First Order Reaction Is 45.6 Sec. What Is The Rate Constant  $k$ ? A)  $1.5 \times 10^{-2} \text{ s}^{-1}$  B)  $1.5 \times 10^{-3} \text{ s}^{-1}$  C)  $1.5 \times 10^{-4} \text{ s}^{-1}$  D)  $1.5 \times 10^{-5} \text{ s}^{-1}$  E)  $1.5 \times 10^{-6} \text{ s}^{-1}$

Appendix B: Agarose Physical Chemistry  
Appendix B: Agarose Physical Chemistry Continued  
The Energy Needed To Melt An Agarose Gel Increases As The Gel Concentration Increases.

The Gelling Temperature Of An Agarose Gel Is Also Influenced By The Mar 1th, 2024

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