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$4 G = 6.67 \times 10^{-11} \text{ N}\cdot\text{m}^2\cdot\text{kg}^{-2}$ $\epsilon_0 = 8.85 \times 10^{-12} \text{ N}\cdot\text{m}^2\cdot\text{C}^{-2}$ $E = 1.6 \times 10^{-19} \text{ C}$

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Number, $23 \cdot 10^{23} \text{ N O}$ 6.02 10 Mol Universal Gas Constant, $R 8.31 \text{ J (mol K)}$

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12 -- W 1:00 P.m. -- 3:45 P.m. 13 W 4:00 P.m. -- 6:45 P.m. ... Person & Remote)
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