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$4 G = 6.67 \times 10^{-11} \text{ N}\cdot\text{m}^2\cdot\text{kg}^{-2}$ $\epsilon_0 = 8.85 \times 10^{-12} \text{ N}\cdot\text{m}^2\cdot\text{C}^{-2}$ $E = 1.6 \times 10^{-19} \text{ C}$
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Number, 6.02×10^{23} Mol Universal Gas Constant, $R = 8.31$ J (mol K)

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