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B) 0.260 C) 0.200 D) 0.225 E) 0.245 16. What Is The
Half Life In Sec For A First Order Reaction With $K =$
 $2.03 \times 10^{-3} \text{ s}^{-1}$? A) 323 B) 369 C) 341 D) 422 E) 395
17. The Half Life Of A First Order Reaction Is 45.6 Sec.
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Appendix B: Agarose Physical Chemistry Agarose
Physical ... Agarose Physical Chemistry Appendix B:
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