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 1.1 Rate Of Reaction And Dependence On Temperature We Will Once Again Look At
 The Formation Of Ammonia (NH_3) From Nitrogen And Hydrogen (see Section
 Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction Follows The
 Equation: $\text{N}_2 + 3\text{H}_2 \rightleftharpoons 2\text{NH}_3$ (1) $\Delta H^\circ = -92 \text{ kJ mol}^{-1}$ $\Delta S^\circ = -192 \text{ J mol}^{-1} \text{ K}^{-1}$ 14th,
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