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Structures)! (only Show Valence Electrons For A Lewis  
Dot Structure)! Both Atoms Now Have 8 Electron 14th,  
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The T-butyl Side. • It Is Better To Ma 13th,  
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Name (last ...C) 1,1-dibromo-2-chloro-2-fluoropropane  
D)1,1-dibromo-2,2 Dichlorofluoropropane 12 Ethyne Is  
A(n) A) Alkane B) Alkene C) Alcohol D) Aromatic E)  
Alkyne 13 How Many Alkyne Isomers Are Possible With  
The Molecular Formula C<sub>6</sub>H<sub>10</sub>? A) 3 Or Less B) 4 C) 5  
D) 6 E) 7 Or More 14 Alkanes Have The General  
Formula A) C<sub>n</sub>H<sub>2n+2</sub> B) C<sub>n</sub>H<sub>2n</sub> C) C<sub>n</sub>H<sub>2n-2</sub> D) C<sub>n</sub>H<sub>n</sub> E  
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Same Two Chemicals In Two Different Ways. The  
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