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What Is The Percentage Weight By Weight Concentration Of Silver In The Metal?
Solution
 $M_{Ag} = 11.1 \text{ G}$
 $M_{Alloy} = 12.0 \text{ G}$
 $C_{Ag} = \frac{11.1 \text{ G}}{12.0 \text{ G}} \times 100\% = 92.5\% \text{ W/W}$
The Ring Is 92.5 % W/W Silver.
Very Low Concentrations In Studies Of Solutions In The Environment, We Often ...
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