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Answers: Moles And Stoichiometry Practice Problems Answers: Moles And Stoichiometry Practice Problems 1) How Many Moles Of Sodium Atoms Correspond To 1.56 x 10²¹ Atoms Of Sodium? $1.56 \times 10^{21} \text{ atoms Na} \times \frac{1 \text{ mol Na}}{6.022 \times 10^{23} \text{ atoms Na}} = 2.59 \times 10^{-3} \text{ mol Na}$ 236.022 x 10 atoms Na 2) Determine The Mass In Grams Of Each Of The Following: A. 1.35 mol Of Fe 1.35 mol Fe x 55.845 g Fe = 75.4 g Fe 1 mol Fe B. 24.5 mol O Mar 1th, 2024

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