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Molar Volume Of An Ideal Gas ($P = 101.325 \text{ kPa}$, $T = 0 \text{ }^\circ\text{C}$) $V_M = 22.412 \text{ dm}^3 \text{ mol}^{-1}$ (39)
Standard Partial Molar Enthalpy H°_f J mol^{-1} Standard Partial Molar En Apr 7th, 2024

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