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(c) Write The Ionic Equation For The Reaction In Which Lead(II) Iodide Is Formed..... [2] (d) Aqueous Silver Nitrate Produces A Yellow Precipitate With Both Iodide Ions And Carbonate Ions. When Testing An Unknown Solution For Iodide Ions, The Aqueous Silver Nitrate Is Acidified. Explain Why The Aqueous Silver Nitrate Is ...

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(d) Barium Chloride Is An Ionic Compound. Draw A Diagram That Shows The Formula Of The Compound, The Charges On The Ions And Gives The Arrangement Of The Valency Electrons Around The Negative Ion. The Electron Distribution Of A Barium Atom Is 2.8.18. 3th, 2024

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Aqueous Potassium Iodide And Aqueous Acidified
Potassium Manganate(VII) Are Added To Aqueous
Solutions Of Iron(II) Sulfate And Iron(III) Sulfate. Iron(II)
Ions, Fe^{2+} , Are Reducing Agents In Aqueous Solution.
Iron(III) $3+$ ions, Fe^{3+} , Are Oxidising Agents In Aqueous
Solution. Complete The Table. 2th, 2024

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Mass Of Ions Present In A 1000 cm^3 Sample Of Water
From A Sulfur Spring. Ion Present Formula Of Ion Mass
Present In The 1000 cm^3 Sample / Mg Br- 4 Calcium
 Ca^{2+} 44 3th, 2024

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38 Which Statement About A Family Of Organic

Compounds Describes An Homologous Series? All Compounds In The Family Have The Same A Functional Group. B Physical Properties. C Relative Molecular Mass. D Structural Formula. 39 Which Column Describes Ethane And Which Colu 3th, 2024

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