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Worksheet 7 - Ideal Gas Law I. Ideal Gas Law The Findings Of 19th Century Chemists And Physicists, Among Them Avogadro, Gay-Lussac, Boyle And Charles, Are Summarized In The Ideal Gas Law:  $PV = NRT$   $P =$  Pressure  $V =$  Volume  $N =$  Moles Of Gas,  $R =$  Universal Gas Constant  $T =$  Temperature. The Value Of  $R$  Varies With The Mar 7th, 2024

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