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A) $[Co(H 2O) 6] 2+ Co +2 E) [FeO 4] 2- Fe +6 I) K 2[CoCl 4] Co +2 B) [CrCl 6] 3- Cr +3 F) [Mn(CN) 6] 4- Mn +2 J) K 3[AuF 6] Au +3 C) [Co(NH 3) 6]Cl 2 Co +2 G) [Ni(CO) 4] Ni 0 K) (NH 4) 2[IrCl 6] Ir +4 D) [Co(NH 3) 5Cl]Cl 2 Co +3 H) [Ni(EDTA)] 2- Ni +2 L) Na[Mn(CO) 5] Mn -1 TASK 2 - Writing Half Equations A) <math>2I- \rightarrow I 2 + 2e - B ...$ Jun 24th, 2024

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Metal Carbonate → Metal Oxide + Carbon Dioxide (on Heating) CaCO 3 → CaO + CO 2 TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. Jan 7th, 2024

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