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We Can Assume The Ideal Gas Equation Of State.  $V = RT/P \Rightarrow (\partial V/\partial T)_P = R/P$ .  
 Substitution  $\Rightarrow \Delta S = C_p \ln(T_2/T_1) - R \ln(P_2/P_1)$  Applying  $\Delta S = 0 \Rightarrow \ln(T_2/T_1) = R/C_p \ln(P_2/P_1) \Rightarrow (T_2/T_1) = (P_2/P_1)^{C_p/C_p - R}$   
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