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We Can Assume The Ideal Gas Equation Of State.  $V = RT/P \Rightarrow (\partial V / \partial T)_P = R/P$ .  
 Substitution  $\Rightarrow \Delta S = C_P \ln(T_2/T_1) - R \ln(P_2/P_1)$  Applying  $\Delta S = 0 \Rightarrow \ln(T_2/T_1) = R/C_P \ln(P_2/P_1) \Rightarrow (T_2/T_1) = (P_2/P_1)^{(C_P - C_V)/C_P}$  Jan 7th, 2024

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