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	100	150	200	250	300	350	450	550	700	1,000	1,100	1,200	1,300
Nitrogen (N ₂)													
Oxygen (Feb 3th, 2024 FUELS AND COMBUSTION 3.1 Introduction To Combustion 87													

The Total Number Of Moles, N, Is The Sum Of The Number Of Moles Of All The Components Of The Mixture: $N = N_1 + N_2 + \dots$ It Follows That The Sum Of All The Mole Fractions Of The Mixture Must Also Equal 1. $X_1 + X_2 + \dots = 1$ The Mass Of Component I In A Mixture Is The Product Of The Number Of Moles Of I And Its Molecular Weight, M_i . The Mass Of The Mixture Is Therefore The Sum, $M = N_1 M_1 + N_2 M_2 + \dots$ Jan 7th, 2024 POLYCOPIE Moteurs à Combustion Interne, Combustion Et ... Vue Globale Du Moteur Volumétrique. Vu De Profil, Le Moteur Possède Un Côté Chaîne De Distribution Et Un Côté Volant Moteur. Les Organes Annexes Au Moteur

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