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[1] (b) A Sample Of Vanadium Chloride Was Weighed  
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(d) The Oxidation States Of Vanadium In Its  
Compounds Are V(+5), V(+4), V(+3) And V(+2). The  
Vanadium(III) Ion Can Behave As A Reductant Or An  
Oxidant. (i) Indicate On The Following Equation Which  
Reactant Is The Oxidant.  $2V^{3+} + Zn \rightarrow 2V^{2+} + Zn^{2+}$   
[1] (ii) Which Change In The Following Mar 1th, 2024

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