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33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$ Actual yield = 1.45g Theoretical yield = 5.95g $\text{KClO}_4 \times 1\text{mol} \times 122.55\text{g} \times 2 \times 16 = 3.11\text{g O}_2$
%Yield = $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$

34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$ Percent purity = 62.8%
pure FeCO_3 May 10th, 2024
Chapter 3 Stoichiometry - Chemistry The Compound Para-aminobenzoic Acid (you May Have Seen It Listed As PABA On Your Bottle Of Sunscreen) Is Composed Of Carbon (61.31%), Hydrogen (5.14%), Nitrogen (10.21%), And Oxygen (23.33%). Find The ... - In Other Words It's The Amount Of Product Possible From Stoichiometry. The "perfect Jan 12th, 2024
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