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Answer Key 1. The Oxidation Number Of A Monatomic  
Ion Is Equal In Magnitude And Sign To Its Ionic Charge.  
For Example, Sodium's Charge Is +1, So Its Oxidation  
Number Is +1. 2. The Oxidation Number Of Hydrogen  
In A Compound Is +1, Except In Metal Hydrides Such  
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Oxidation With Nickel Peroxide. I. Oxidation Of Alcohols Nickel Peroxide In An Aqueous Alkaline Solution Afforded The Corresponding Carboxylic Acids, While Carbonyl Compounds Were Obtained On Oxidation Of Primary Or Secondary Alcohols By This Reagent In Organic Solvents. Although Such Metal

Oxides As Manganese Di- Oxide, Selenium Dioxide, Lead Dioxide, Silver Oxide, Jan 2th, 2024  
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