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Total Moles Of Gas = 1.3 Mol + 1.5 Mol = 2.8 Mol . P = (2.8mol) X (0.0821 L-atm Mol-K) X (273K) 10 L = 6.3 Atm Or 4.8 X 103 Torr. Title: WORKSHEET 16 Name \_\_\_\_\_ Autho 19th, 2024

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- The Burner Must Not Be Used Outside Of Its Capacity Range (see Capacity Graphs, Ch. 8.2).
- The Gas Supply Pressure Must Not Exceed The Gas Pressure Given On The Burner Plate.
- The Burner Is Not Preset. Burner Type Forced Draught Burners With Two Sta 4th, 2024

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Worksheet 7 - Ideal Gas Law I. Ideal Gas Law The Findings Of 19th Century Chemists And Physicists, Among Them Avogadro, Gay-Lussac, Boyle And Charles, Are Summarized In The Ideal Gas Law:  $PV = NRT$  P = Pressure V = Volume N= Moles Of Gas, R = Universal Gas Constant T = Temperature. The Value Of R Varies With The 7th, 2024

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