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9-1 (a) A Weak Electrolyte Only Partially Ionizes When
Dissolved In Water. NaHCO_3 Is An Example Of A Weak
Electrolyte. (b) A Brønsted-Lowry Acid Is A Molecule
That Donates A Proton When It Encounters A Base
(proton Acceptor). By This Definition, NH_4^+ Can Be A
Brønsted-Lowry Acid. (c) The Conjugate Acid Of A
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