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The Purpose Of The Laboratory Is To Experimentally Determine How Temperature Influence The Solubility Equilibria Of A Salt With Low Solubility, Ca(OH) 2. An Experimental Determination Of The Solubility Product Constant, K Sp, At Two Temperatures Will Be Employed To Determine H<sup>o</sup>,  $\Delta$ S<sup>o</sup>,  $\Delta$  And  $\Delta$ G<sup>o</sup> For The Solubility Reaction. INTRODUCTION. Most Hydroxides (except Those Alkali Metals And ... 6th, 2024

# Dependence Of The Solubility On Temperature

M;i Is The Partial Molar Enthalpy And S M;i Is The Partial Molar Enthropy Of The I-th Component. Substi-tuting This Into Eq. 3, We Get The Correlation Lnx Saturated = H0 Diss H M,solid R! 1 T + S0 Diss S M,solid R (6) This Includes Partial Molar Quantities, Values That May Also Be Composi 8th, 2024

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