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(3) The Compressibility Factor,  $Z$ , Is Defined As,  $Z := \frac{Pv}{RT} = \frac{P}{P^*} \rightarrow \frac{RT}{P^*v}$ . (3.81) If  $Z \uparrow 1$  For A Gas, Then It Can Be Modeled Well With The Ideal Gas Model. The Compressibility Factor,  $Z$ , Is Plotted In Figure 3.27 For A Variety Of Substances As A Function Of The Reduced Pressure,  $P/P^*$ , And Reduced Temperature,  $T/T^*$ , where  $P^*$  And  $T^*$  Jan 2th, 2024

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