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 $+332 \text{ J K}^{-1} \text{ Mol}^{-1}$  B  $+189 \text{ J K}^{-1} \text{ Mol}^{-1}$  C  $+145 \text{ J K}^{-1} \text{ Mol}^{-1}$  D  $+85 \text{ J K}^{-1} \text{ Mol}^{-1}$  (Total 1 Mark) Q25.Which One Of The Equations Below Represents A Reaction That Is Feasible ... Apr 13th, 2024

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