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6] $2+$ Co +2 E) [FeO 4] $2-$ Fe +6 I) K 2 [CoCl 4] Co +2 B) [CrCl 6] $3-$ Cr +3 F)
[Mn(CN) 6] $4-$ Mn +2 J) K 3 [AuF 6] Au +3 C) [Co(NH 3) 6]Cl 2 Co +2 G) [Ni(CO) 4] Ni
0 K) (NH 4) 2 [IrCl 6] Ir +4 D) [Co(NH 3) 5Cl]Cl 2 Co +3 H) [Ni(EDTA)] $2-$ Ni +2 L)
Na[Mn(CO) 5] Mn -1 TASK 2 – Writing Half Equations A) $2I^- \rightarrow I_2 + 2e^-$ – B ... Jan 7th,
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It Burns ... Jan 23th, 2024© Wwww.CHEMSHEETS.co.uk 20-May-12 Chemsheets AS
008 1Metal Carbonate \rightarrow Metal Oxide + Carbon Dioxide (on Heating) $CaCO_3 \rightarrow CaO$
+ CO_2 TASK 4 – WRITING BALANCED EQUATIONS 1) Balance The Following
Equations. Jan 22th, 2024.
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