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33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$  Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1\text{mol} \times 122.55\text{g} \times 2 = 245.10\text{g}$   $\text{KClO}_4 \times 32.00\text{g} \times 1\text{mol} = 32.00\text{g}$

$\% \text{Yield} = \frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$  Percent purity = 62.8%?  $\text{pure FeCO}_3$  Feb

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Oxygen ( $\text{O}_2$ ).  $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$  C. Jan 3th, 2024 Chapter 11 Study Guide Chemistry Stoichiometry Answer Key Chapter 11

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Name: UNIT 8: Acid And Base Chemistry UNIT TEST REVIEW ... 1. Which Two Formulas Represent Arrhenius Acids? (1) CH<sub>3</sub>COOH And CH<sub>3</sub>CH<sub>2</sub>OH (2) HC<sub>2</sub>H<sub>3</sub>O<sub>2</sub> And H<sub>3</sub>PO<sub>4</sub> (3) KHCO<sub>3</sub> And KHSO<sub>4</sub> (4) NaSCN And Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> 2. The Only Positive Ion Found In H<sub>2</sub>SO<sub>4</sub> (aq) Is The Ammonium Ion (2) Hydronium Ion (3) Hydroxide Ion (4) Sulfate Ion PH Scale Scale: HpH May 11th, 2024

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