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Environmental Chemistry 2.1 INTRODUCTION 2.2 STOICHIOMETRY ...42 Chapter 2 Environmental Chemistry Section 2.2 Stoichiometry 43 We Already Know That There Are 44 Grams Per Mole Of CO" So We Do Not Need To Recalculate That. Two Moles Of Butane (2 Mol X 58 G/mol = 116 G) Yields 8 Moles Of Carbon Dioxide (8 Mol X 44 G/mol = 352 G CO₂) , So We Can Set Up The Following Proportion: Mar 1th, 2024.

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www.apchemistrynmsi.wikispaces.com 6 Exercise 7 Molar Mass And Numbers Of Molecules Isopentyl Acetate (C₇H₁₄O₂), The Compound Responsible For The Scent Of Bananas, Can Be Produced Commercially. Interestingly, Bees Release About 1μg (1 × 10⁻⁶ G) Of This Compound When They Sting. The Resulting Scent Attracts Apr 1th, 2024
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Name: ___KEY___ 33. $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$
Actual yield = 1.45g Theoretical yield = 5.95g
 $\text{KClO}_4 \times 1 \text{ mol} = 122.55 \text{ g}$
 $2\text{O}_2 \times 1 \text{ mol} = 64.00 \text{ g}$
 $\text{KClO}_4 \times 32.00 \text{ g} = 3.11 \text{ g O}_2$
 $\% \text{ Yield} = \frac{1.45 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$
34. $4\text{FeCO}_3 + \text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$
Percent purity = 62.8%
pure FeCO_3 Mar 1th, 2024

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