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 You Must Memorize The Conversion (for Example: 1000  
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 Remember That Metric Conversions Are Exact Ratios  
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 Complete Each Of The Following Conversions To The  
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Mole Conversion Practice ProblemsMole Conversion  
 Chart Grams → Moles → Particles 1.62 G CaCl 2 × 1 Mol  
 1g ( \ | \ ) | × 6.022×10<sup>23</sup> Particles Mol ( \ | \ ) |  
 Practice Problems: 1. A Sample Of Neon Has A Volume  
 Of 75.8 L At STP . Calculate The Number Of Moles. 3.38  
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