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Atoms Held Together By Chemical Bonds. Example: Table Salt, NaCl Also The Smallest Unit Of A Compound. • Not All Molecules Are Compounds. H<sub>2</sub>, O<sub>2</sub>, And Other Diatomic Gases Are Not Molecules. Why? 3 How Are Molecules Formed? • The Number Of Electrons In The Outermost Electron Shell Determine Whether An Atom Is Reactive Or Inert. Carbon ... 13th, 2024

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• Neutrons (neutrally Charged Particles) 2) Electrons (negatively Charged Particles) • Orbit The Nucleus An Atom Is Electrically Neutral (protons = Electrons)

Chapter 2: Atoms, Molecules & Life Figure 2.1

-Audesirk2 & Byers Periodic Table Of Elements: Element = Substance That Can't 14th, 2024

## **The History & Make-Up Of Atoms Atoms**

For Each Element, We Can Directly Relate The Amount Of Protons, Electrons And Neutrons That Exist. But First, We Need To Learn Some Terms! Symbol Of Element 1 Or 2 Letter Abbreviation For Each Element Mass Number • Not Always A Whole Number (more On This Later!) • #protons + #neutrons Atomic Number #protons (defines The Element!) Thus, 6th, 2024

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## **Periodic Table**

Atoms And Bonding Guided Reading And Study Atoms, Bonding, And The Periodic Table (continued) Valence Electrons And Bonding 1. \_\_\_\_\_ Are Those Electrons That Are Held Most Loosely In An Atom. 2. Is The Following Sentence True Or False? The Number Of Valence Electrons In An Atom Of An Element Determines The Ways In Which The Atom Can 12th, 2024

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Element. Neutral Atoms Have The Same Number Of Electrons As Protons. Therefore, The Atomic Number Of An Element Equals The Number Of Electrons In An Atom Of The Element. EHHDBG@ K 3. Infer The Nucleus Of A Helium Atom Contains Four Subatomic Particles. How Many Neutrons Does The Helium Atom Have? READING CHECK 4. Describe Why Are Atoms ... 9th, 2024

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When Two Atoms Approach Each Other, These External Electrons Interact. The Electrons Reject, But Are Attracted To The Protons Within The Atoms. The Interaction Of Forces Causes Some Atoms That Form Bonds With Each Other And Attack Together. The Two Main Types Of Bonds Formed Between Atoms Are Ionic

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**Moles Atoms To # Atoms 1,2. A) B) C) D) E) AB)  
A) B) C) D) AB)**

A)  $5.679 \times 10^{-22}$  G F B)  $1.136 \times 10^{-21}$  G F C)  $2.840 \times 10^{-22}$  G F D)  $2.989 \times 10^{-23}$  G F E)  $3.850 \times 10^{26}$  G F  
AB) None Of These 18 Atoms F X 1 Mol F  $6.022 \times 10^{23}$   
Atoms 10th, 2024

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6 © 2014 Pearson Education, Inc. How Many Neutrons Does Li Have? 1. 3 2. 4 3. 7 4. 10 3 4 7 10 25% 25%  
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Shannan Maisey S.maisey@unsw.edu.au Teaching Support Officer  
Trinah De Leon Trinah@unsw.edu.au 9385 4651 Lab Coordinator  
Ron Haines R.haines@unsw.edu.au Lecturer (weeks 1-4)  
Tim Schmidt Timothy.schmidty@unsw.edu.au Lecturer (weeks 5, 7-10)  
Junming Ho Junming.ho@unsw.edu.au 14th, 2024

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A) Microwave

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