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10 When Vanadium(II) Compounds Are Dissolved In
Water, The Following Equilibrium Is Established. $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O + OH^-$ What Would Alter The
Composition Of The Equilibrium Mixture In Favour Of
The V^{2+} Ions? A Adding An Acid B Adding A Reagent
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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use The Following Data To Calculate A Value For The Lattice Energy Of AgBr(s).
First Ionisation Energy Of Silver = +731kJmol⁻¹ Electron Affinity Of Bromine = 325kJmol⁻¹ Enthalpy Change Of Atomisation Of Silver = +285kJmol⁻¹ Enthalpy Change Of Atomisation Of Bromine = +112kJmol⁻¹ ... 1th, 2024

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