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CHEMISTRY 9701/01 1 Hour 10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established.  $V^{2+} + H_2O \rightleftharpoons V^{3+} + H_2O$  What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The  $V^{2+}$  Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitates Jan 2th, 2024

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