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4. $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$ 5. $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$

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Whether Or Not A Single-replacement Reaction Will Occur Can Be Determined By Using An Activity Series. The Activity Series Of Metals Lists Metals In Order Of

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The Reactants Are The Elements That React With One And Another In A Chemical Reaction, While The Products Are The Elements That We Get After The Reaction. The Chemical Equation Has The Products On The Right Side, While The Reactants Are Written On The Left Side. Both Of Them Are Separated By An Arrow. May 4th, 2024

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