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Classify Each Of The Reactions In Problems 1–7 As To Type. 8. Use The Activity Series Of Metals (Table 8.2) And Your Knowledge Of The Relative Reactivity Of The Halogens To Predict Whether The Following Reactions Will Occur. Write Balanced Equations For Those Reactions That Do Occur. A.  $\text{Br}_2(l) + 1 \text{ NaCl}(aq) \rightarrow$  Y B.  $\text{Ca}(s) + 1 \text{ Mg}(\text{NO}_3)_2(aq) \rightarrow$  Y C. K ... Jun 1th, 2024

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Name: \_\_\_\_\_ Suggested Answers Date: \_\_\_\_\_ Predicting Products Of Chemical Reactions - Practice Problems Directions: Predict The Products For, And Then Balance Each Of The Following Chemical Reactions: 1.  $\text{SiI}_4 + \text{Mg}$  (single Replacement)  $\text{SiI}_4 + 2\text{Mg} \rightarrow 2\text{MgI}_2 + \text{Si}$  2.  $2\text{Al} + 3\text{I}_2$  (synthesis)  $2\text{Al} + 3\text{I}_2 \rightarrow 2\text{AlI}_3$  3.  $\text{Cu} + \text{C}$  Feb 1th, 2024

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4.  $\text{HCl(aq)} + \text{Sr(OH)}_2\text{(aq)} \rightarrow \text{SrCl}_2\text{(aq)} + \text{H}_2\text{O(l)}$  5.  $\text{AlBr}_3\text{(s)} \rightarrow \text{Al(s)} + \text{Br}_2\text{(l)}$  Classification Of Chemical Reactions Reactions, Much Like Elements And Compounds, Can Be Classified According To Type. The Ability To Recognize And Classify Reactions Can Help Us Predict Feb 1th, 2024

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