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1 Chemical

Reactions 1.1 Rate Of Reaction And Dependence On Temperature
We Will Once

Again Look At The Formation Of Ammonia (NH₃) From Nitrogen And Hydrogen (see

Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction

Follows The Equation: $N_2 + 3H_2 \rightleftharpoons 2NH_3$ (1) $H_0 = 92 \text{ KJ Mol}^{-1}$ $S_0 = 192 \text{ J Mol}^{-1} \text{ K}^{-1}$

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Grading: Exam 1 15%

Exam 2 15% Exam 3 15% Final Exam 20% Research Paper 10% Homework 20%

Class Participation 5% Last Year, The Following Grading Scale Was Used: A ≥ 92 A/B

≥ 89 B ≥ 82 B/C ≥ 80 C ≥ 71 C/D ≥ 69 D ≥ 61 F ≤ 60 If Necessary, The Grading

Scale May Be Curved Downward. 1th, 2024.

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Solution Manual [194]. Laser Electronics Solution Manual Joseph T. Verdeyen 1994
[195].GNU Octave Is Software Featuring A High-level Programming Language,
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