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Again Look At The Formation Of Ammonia ( $\text{NH}_3$ ) From Nitrogen And Hydrogen (see  
Section Chemical Equilibrium Of The Thermodynamics Chapter). This Reaction

Follows The Equation:  $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$  (1)  $H_0 = 92 \text{ kJ/mol}$   $S_0 = 192 \text{ J/mol K}$  4th,

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$\geq 89$   $B \geq 82$   $B/C \geq 80$   $C \geq 71$   $C/D \geq 69$   $D \geq 61$   $F \leq 60$  If Necessary, The Grading

Scale May Be Curved Downward. 1th, 2024.

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[195].GNU Octave Is Software Featuring A High-level Programming Language,  
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