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(chapter 2.5 And 2.6) 1. No One Resonance Forms Accurately Depicts The Structure Of The Molecule. The Real Structure Is A Composite Or Hybrid Of All Resonance Forms 2. Resonance Forms Differ Only By The Placement Of π - Or Non-bonding

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Covalent Bonds Form Through The Process Of Sharing Of Electrons. Two Atoms Can Fill Their Outer Shells By Sharing Electrons. In Fact, Atoms Give Up Little By Sharing. For Example, 2 Hydrogen Atoms Share Their Electron To Have Two Electrons In Their Shell. Oxygen ($1s^2, 2s^2, 2p^4$) Shares Two Jan 18th, 2024

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