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Answers: Moles And Stoichiometry Practice ProblemsAnswers: Moles And

Stoichiometry Practice Problems 1) How Many Moles Of Sodium Atoms Correspond To  $1.56 \times 10^{21}$  Atoms Of Sodium?  $1.56 \times 10^{21} \text{ Atoms Na} \times \frac{1 \text{ Mol Na}}{6.022 \times 10^{23} \text{ Atoms Na}} = 2.59 \times 10^{-3} \text{ Mol Na}$

2) Determine The Mass In Grams Of Each Of The Following: A. 1.35 Mol Of Fe  $1.35 \text{ Mol Fe} \times 55.845 \text{ G Fe} = 75.4 \text{ G Fe}$  B. 24.5 Mol O  $24.5 \text{ Mol O} \times 16.00 \text{ G O} = 392 \text{ G O}$

Balance The Following Chemical Reactions: A.  $2 \text{ CO} + \text{O}_2 \rightarrow 2 \text{ CO}_2$  B.  $2 \text{ KNO}_3 \rightarrow 2 \text{ KNO}_2 + \text{O}_2$  C.  $2 \text{ O}_3 \rightarrow 3 \text{ O}_2$  D.  $\text{NH}_4\text{NO}_3 \rightarrow \text{N}_2\text{O} + \dots$

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The Equation First Then Do Stoichiometry As Described In The Question. 7:30 PM - A ... Feb 3th, 2024  
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 Stoichiometry Practice Problems  
 Stoichiometry Practice Problems \*\*Balance The Following Equations FIRST, Then Answer The Questions: Mole-to-Mole: 1.  $N_2 + H_2 \rightarrow NH_3$  How Many Moles Of Hydrogen Are Needed To Completely React With 2.0 Moles Of Nitrogen? 2.  $C_3H_8 + O_2 \rightarrow CO_2 + H_2O$  How Many Moles Of Oxygen Are Neces Jan 3th, 2024.

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