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Chapter 8: Ionic And Metallic Bonding 8.1 Ions Practice 8.1 Write The Ground State Configuration For The Metal Calcium, And Predict The Ion Form It Must Become To Be Isoelectric With The Nearest Noble Gas. Answer: Ca: $1s^2 2s^2 2p^6 3s^2 3p^4 4s^2$ Or $[Ar]4s^2$ The Nearest Noble Gas To Calcium Is $[Ar]$. So Calcium Will Typically Lose Two Electrons To Become Jan 9th, 2024

Chemistry Student Edition - Basic Answer Key Chapter 9 ...

Electrons In Different Bonding Pairs Causes Them To Remain As Far Apart As Possible. 2. A. 4 B. 6 C. 3 D. 2 3. Molecular Geometry Only Considers Atoms, While Electron Domain Geometry Includes Lone Pairs 4. Lone Pairs Take Up More Space Than Bonding Pairs 5. A. Bent,