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2. Q. List Three Subatomic Particles Of An Atom. Compare Them On The Basis Of Relative Mass And Charge In A Tabular Form. Ans: Particle Symbol Charge Mass Electron E---1.60×1019 C 9.1×10-31 Kg Proton +p (H+) 1.60×10-1 10th, 2024

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(c) Three Mole (d) Four Moles (4) Which Of The Following Weighs The Most? (a) One G-atom Of Nitrogen (b) One Mole Of Water (c) One Mole Of Sodium (d) One Molecule Of H2So4 (5) 5.0 Litre Of 11th, 2024

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