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 1. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 2. $2\text{Na} + \text{Cl}_2 \rightarrow 2\text{NaCl}$ 3. $\text{N}_2 + \text{O}_2 \rightarrow 2\text{NO}$ 4. $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$ 5. $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$
 + O_2 6. $3\text{Ca} + \text{N}_2 \rightarrow \text{Ca}_3\text{N}_2$ 7. $2\text{Li} + \text{F}_2 \rightarrow 2\text{LiF}$ 8.
 $3\text{Mg} + \text{N}_2 \rightarrow \text{Mg}_3\text{N}_2$ 9. $2\text{NH}_3 \rightarrow \text{N}_2 + 3\text{H}_2$ 10.
 $2\text{HCl} \rightarrow \text{H}_2 + \text{Cl}_2$... Feb 4th, 2024 Balancing Chemical
 Equations - Answer Key Balancing Chemical Equations
 - Answer Key Balance The Equations Below: 1) $1\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$ 2) $2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$ 3) $2\text{NaCl} + \text{F}_2 \rightarrow 2\text{NaF} + \text{Cl}_2$ 4) $2\text{H}_2 + \text{O}_2 \rightarrow 2\text{H}_2\text{O}$ 5) $1\text{Pb}(\text{OH})_2 + 2\text{HCl} \rightarrow 2\text{H}_2\text{O} + 1\text{PbCl}_2$ 6) $2\text{AlBr}_3 + 3\text{K}_2\text{SO}_4 \rightarrow$

$\text{KBr} + 1 \text{ Al}_2 (\text{SO}_4)_3 + 7 \text{ H}_2 \text{O} \rightarrow 1 \text{ CH}_4 + 2 \text{ O}_2 + 1 \text{ CO}_2 + 2 \text{ H}_2$

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