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Acetanilide stand in a one-to-one molar relationship, so 0.0322 mol of aniline produces 0.0322 mol of acetanilide. The amount of acetic anhydride does not play a role, since it is in excess. Then, $n(\text{aniline}) = n(\text{Acetanilide}) = 0.0322$. We find the molar mass of acetanilide: $M(\text{Acetanilide}) = 12 \times 8 + 1 \times 9 + 14 \times 1 + 16 \times 1 + 135 = 179$. 4th, 2024 Chemistry MCQs Chemistry Multiple Choice Questions Answers ... Chemistry MCQs Chemistry Multiple Choice Questions Answers 1/8 Downloaded From Holychild.org On March 7, 2021 By Guest Kindle File Format Chemistry MCQs Chemistry Multiple Choice Questions Answers When somebody should go to the ebook stores, search creation by shop, shelf by shelf, it is in point of fact problematic. 1th, 2024 Chemistry MCQs Chemistry Questions BHU, IP MBBS, JIPMER, State PMT, MHT CET, you can solve these question papers online. Chemistry MCQs | Chemistry Questions Chemistry is the branch of science which deals with the compounds composed of atoms, i.e. elements, and molecules, i.e. combinations of atoms: their composition. 2th, 2024.

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