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TABLE OF CONJUGATE ACID-BASE PAIRS Acid Base K A (25 OC) HClO 4 ClO 4 - H 2 SO 4 HSO 4 - HCl Cl- HNO 3 NO 3 - H 3 O + H 2 O H 2 CrO 4 HCrO 4 - 1.8 X 10<sup>-1</sup> H 2 C 2 O 4 (oxalic Acid) May 3th, 2024

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Chapter 3 Acid-Base Equilibria Acid-Base Equilibria Acids And Bases Play A Key Role In A Number Of Environmentally

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### **(aq) Acid Base Conjugate Conjugate Acid Base**

Acid Base Conjugate Conjugate . Acid Base . 2) What Is The Strongest Base In The Following Reaction?  $\text{HNO}_3(\text{aq}) + \text{H}_2\text{O}(\text{l}) \rightleftharpoons \text{NO}_3^-(\text{aq}) + \text{H}_3\text{O}^+(\text{aq})$   $\text{H}_2\text{O}$  Is The Strongest Base. Strong Acids, Such As  $\text{HNO}_3$  Have Weak Conjugate Bases, So  $\text{NO}_3^-$  is A Weak Base. H<sub>2</sub>O And May 4th, 2024

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