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3, Benzenesulfonic Acid Is Produced.  $\text{SO}_3\text{H}$  Benzenesulfonic Acid +  $\text{SO}_3$  The Mechanism Of This Reaction Is Similar To That Of The Nitration Of Benzene. Concentrated  $\text{H}_2\text{SO}_4$  Is Used In An Initial Step To Generate The  $\text{SO}_3\text{H}^+$  Electrophile As Shown.  $\text{SO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{SO}_3\text{H}^+ + \text{HSO}_4^-$  (i) Draw A Mechanism For The Reaction Of Benzene With  $\text{SO}_3\text{H}^+$  ... 3th, 2024

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10 When Vanadium(II) Compounds Are Dissolved In Water, The Following Equilibrium Is Established. V<sup>2+</sup> + H<sub>2</sub>O V<sup>3+</sup> + H<sub>2</sub>O + OH<sup>-</sup> What Would Alter The Composition Of The Equilibrium Mixture In Favour Of The V<sup>2+</sup> Ions? A Adding An Acid B Adding A Reagent That Selectively Precipitat 2th, 2024

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