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3, Benzenesulfonic Acid Is Produced. SO_3H Benzenesulfonic Acid + SO_3 The Mechanism Of This Reaction Is Similar To That Of The Nitration Of Benzene. Concentrated H_2SO_4 Is Used In An Initial Step To Generate The SO_3H^+ Electrophile As Shown. $\text{SO}_3 + \text{H}_2\text{SO}_4 \rightarrow \text{SO}_3\text{H}^+ + \text{HSO}_4^-$ (i) Draw A Mechanism For The Reaction Of Benzene With SO_3H^+ ... Feb 3th, 2024

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MATH 235/W08: Orthogonal Diagonalization, Symmetric ...

2. Find A Unitary Diagonalization Of The Following Skew-symmetric Matrix $A = \begin{pmatrix} 0 & 1 & -1 & -1 \\ 1 & 0 & 1 & 1 \\ -1 & 1 & 0 & 1 \\ 1 & -1 & -1 & 0 \end{pmatrix}$. BEGIN SOLUTION: As Above, In Problem 7, We Note That The Sum Along Each Row Is 0. Therefore $\lambda_1 = 0$ Is An Eigenvalue W Jun 4th, 2024

221 W08 52 Lect 05 Post - Faculty.concordia.ca

3.1 Molecular Formula & Degree Of Unsaturation "unsaturated" Hydrocarbons: Do Not Have Max. # H's Per C Missing 2H For Every 1 π -bond Or Ring An "element Of Unsaturation" = A π -bond Or A Ring To Help Deduce Structure From Fo Apr 2th, 2024

W08-Gathering Agenda

3. Ask The Class To Take 5 Minutes To Identify At Least 3 Strengths And 3 Weaknesses When It Comes To Their Roles As An Online Student. (This Could Be A Specific Subject, A Skill, Learning A New Language, A Habit, ability, Etc.) 4. After 5 Minutes, Invite The Students To Look At The May 4th, 2024

205 W08 03 Lecture 05 - Concordia University

2.6 The Periodic Table: Organized According To Both atomic Number (Z) & Properties Periods Or Rows (horizontal): Z Increases Left \rightarrow right (8) Elements With Similar Properties are In Vertical "groups" Groups In Red Boxes: 1A - 8A "main Group" Elements 1 Groups In Blue Box: 3B - 8B, 1B, 2B "tran May 1th, 2024

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2. Molecular Orbital (MO) Theory (aka Delocalized Electron Model) Robert S. Mulliken Bonding E-sspread Out Over Entire Molecule, In MOLECULAR ORBITALS Our Focus In Chem 205. (4) 10.2: VB Theory Part (a): Orbital Overlap Model Of Bonding

IN GENERAL: Two Overlapping Orbitals + 2e= One Mar 2th, 2024

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